## PHARMCON freeCE Monograph

Overview of Mouthwashes


Peter Kreckel, RPh
Retired Adjunct Assistant Professor,
Saint Francis University

In the 1930s, scientists examined the relationship between tooth decay in children and naturally occurring fluoride in drinking water. In 1945, the city of Grand Rapids, Michigan was the first to add fluoride to its city water system to provide residents with its benefits. Community water fluoridation has been identified as the most costeffective method of delivering fluoride to all, reducing tooth decay by $25 \%$ in children and adults. Because of its contribution to the dramatic decline in tooth decay over the past 75 years, CDC named community water fluoridation as 1 of 10 great public health achievements of the $20^{\text {th }}$ century. ${ }^{1}$

Over 90 years of research has gone into providing fluoride to the masses, but this practice is controversial in some parts. The community of Tyrone, where my wife, Denise, and I lived for 36 years, had one such dispute. The Borough Council voted to remove fluoride from the town's water supply. With dentists, pharmacists, and a chemistry professor with a PhD all living in town, they chose to consult a funeral director! At the next meeting, a couple of dentists showed chart data from kids who lived in the borough versus the kids who drank well water. Fortunately, cooler heads prevailed, and the fluoride supplementation was reinstated.

The opposition always quotes the label from the bulk storage of sodium fluoride, which says: "[i]ngestion may cause vomiting, abdominal pain, diarrhea, convulsions, collapse, thirst, disturbed color vision, acute toxic nephritis. SODIUM FLUORIDE reacts with acids to form corrosive and toxic hazard forms. ${ }^{2}$

In the words of Paracelsus: "Dosis sola facit venenum" - 'only the dose makes the poison.'

## Halitosis

Let's briefly discuss halitosis, or bad breath.
The major contributor to halitosis is the presence of volatile sulfur compounds. These compounds arise from the breakdown
of foods, dental plaque, and the anaerobic bacteria associated with oral disease. Highprotein foods, such as eggs, meat, and fish, cause the highest concentration of volatile sulfur compounds when broken down by anaerobic bacteria towards the back of the mouth and throat. These bacteria can live in all the corners of the mouth, especially between teeth and in the "nooks and crannies" of the tongue. (See Figure 1)

Approximately $90 \%$ of the vola le sulfur compounds found in breath are:

- Hydrogen sulfide: paper-mill or rotten egg odor
- Methyl mercaptan: rotten cabbage odor
- Dimethyl sulfide: cabbage odor


## Mouthwashes

There are two distinct types of mouthwash:

- Cosmetic mouthwashes only freshen breath and leave the mouth with a pleasant taste. They don't deal with the root cause by killing the bacteria that cause bad breath or help to reduce plaque, gingivitis, or cavities.
- Therapeutic mouthwashes control plaque and gingivitis. Some examples are chlorhexidine, povidone iodine, cetylpyridinium chloride (CPC) and essential oils. Using these mouthwashes significantly reduced plaque and bacterial indicators of gingivitis. A common counseling point is to rinse for at least 60 seconds, then spit the mouthwash out.


## Ingredient(s):

- Cetylpyridinium chloride is an antiseptic that kills off the offending bacteria. CPC decreases dental plaque and gingivitis without the use of essential oils that cause "medicine breath."
- Chlorhexidine (Peridex) is prescriptiononly. It may cause teeth staining, but this can be easily removed by a dental hygienist.

Figure 1. Causes of halitosis


- Fluoride strengthens teeth for added protection against tooth decay. Fluoride reduces demineralization on by strengthening and protecting the enamel on teeth. The most common compound is sodium fluoride ( NaF ), and its typical over-the-counter concentration is $0.05 \%$.
- Peroxides are typically used as bleaching agents. They help to whiten teeth and remove stains over time. The most common forms are hydrogen peroxide or carbamide peroxide. Products that contain 10\% carbamide peroxide yield approximately $3.5 \%$ hydrogen peroxide. Common side effects of these products are dentin hypersensitivity and gum irritation. Hypersensitivity may be exacerbated through peroxides dissolving away the tooth's protective mucus layer. Alcohol levels can be as high as $20 \%$, so it is best practice to avoid use in patients with alcohol use disorder and adolescents due to potential abuse.

Note: For patients with xerostomia, or dry mouth, use:

- Sugar-free products, which reduce tooth decay risk due to decreased salivary flow
- Alcohol-free dental rinses, which decrease dryness caused by xerostomia and alcohol
- Biotene ${ }^{\circledR}$ or Oasis ${ }^{\circledR}$ mouth rinse


## Fluoride

The American Dental Association, the American Academy of Pediatrics, the US Public Health Service, and the World Health Organization all support the fluoridation of water.

The risk of dental caries, or cavities, is reduced due to the uptake of fluoride by enamel crystallites and formation of fluorohydroxyapatite, which is resistant to acid solubilization. Fluoride is anticariogenic because it replaces the hydroxyl ion in hydroxyapatite with the fluoride ion to form fluorapatite on the outer surface of the enamel. Fluorapatite hardens the enamel and makes it more acid resistant. Fluoride also has demonstrated antibacterial activity. Fluoride is most beneficial from birth to age 12 because unerupted permanent teeth are mineralizing at that time.

Fluoride supplementation that a person receives depends on the concentration of fluoride in the drinking water. Adding fluoride to the water reduces dental caries by $25 \%$.

## Sources:

- Drinking water:
- Approximately $73 \%$ of the US population lives in a municipality with a fluoridated water system - The United States Department of Health \& Human Services recommends an optimal fluoride concentration of 0.7 milligrams/liter ( $\mathrm{mg} / \mathrm{L}$ ). This concentration of fluoride in drinking water is the concentration that provides the best balance of protection from
dental caries, while limiting the risk of excess fluoride (dental fluorosis) ${ }^{1}$
- Where I live, in Blair County, Pennsylvania, we have 30 different water systems. ONLY the Tyrone water system is fluoridated. The major municipalities in our county of Hollidaysburg and Altoona do not fluoridate their water supply.
- Dentist-applied: fluoride-based varnishes and gels can be applied directly to the teeth during a dental visit
- OTC mouthwashes: avoid in children under the age of 6 years old, unless directed by a dentist, because they may swallow large amounts of the liquid inadvertently
- Rx Gels and toothpastes: deliver about four times as much fluoride as the OTC products
- Example: Prevident 5000®


## Does It Work?

The prevalence of dental caries in at least one permanent tooth (excluding wisdom teeth) decreased from $90 \%$ among those aged 12-17 years in the 1960s to $60 \%$ among those aged 12-19 years in 1999-2004.

You can check your water supply, or the water supply where you practice at: https://hccd.cdc.gov/DOH_MWF/Default/Defa ult.aspx

## Patient Care Tips for Fluoride

 Supplements- Fluoride supplements (oral) are prescription-only
- All tablet formulations MUST BE CHEWED OR CRUSHED! It allows fluoride into the enamel of the teeth.
- Fluoride may be ingested from bottled water or juices that have been fluoridated
- Avoid milk and other dairy products due to potential for decreased absorption
- Available as sodium fluoride (Luride ${ }^{\circledR}$ ):
- Chewable tablets:
- 0.25 mg
- 0.5 mg
- 1 mg
- Solution: $0.5 \mathrm{mg} / \mathrm{mL}$
- Sodium fluoride: - $2.2 \mathrm{mg}=1 \mathrm{mg}$ active fluoride ion - $1.1 \mathrm{mg}=0.5 \mathrm{mg}$ active fluoride ion
- Most recommend fluoride be given at bedtime, after kids brush their teeth

Obviously, people should drink municipal water to get the benefits of fluoride! And that includes the astronomical rise in soda consumption...

Have a great day on the bench!!

Table 1. How to supplement patients whose water supply is not fluoridated

| Water concentration of fluoride | Age of patient | Supplementation mg/day |
| :---: | :---: | :---: |
| $>0.6 \mathrm{ppm}$ of fluoride | 6 months -3 years | $0 \mathrm{mg} / \mathrm{day}$ |
| $0.3-0.6 \mathrm{ppm}$ of fluoride | $3-16$ years | $0 \mathrm{mg} / \mathrm{day}$ |
|  | 6 months-3 years | $0 \mathrm{mg} / \mathrm{day}$ |
|  | $3-6$ years | $0.25 \mathrm{mg} / \mathrm{day}$ |
|  | $6-16$ years | $0.5 \mathrm{mg} / \mathrm{day}$ |
|  | 6 months -3 years | $0.25 \mathrm{mg} / \mathrm{day}$ |
|  | $3-6$ years | $0.5 \mathrm{mg} / \mathrm{day}$ |

## References

1. CDC. Water fluoridation basics. Centers for Disease Control and Prevention. Published 2019.
https://www.cdc.gov/fluoridation/basics/index.htm
2. Veiligheidstechnici van. Handling Chemicals Safely 1980.; 1980. 842.

## Test Questions

To receive CE credit, access the monograph and take the test online at freeCE.com

1. Which of the following is a major volatile sulfur compound in halitosis?
a. Carbon dioxide
b. Sodium chloride
c. Hydrogen sulfide
d. All of the above statements are true.
2. Which of the following measurements is the optimal fluoride concentration for water, as listed by the United States Department of Health \& Human Services?
a. $\quad 0.7 \mathrm{mg} / \mathrm{L}$
b. $7 \mathrm{~g} / \mathrm{L}$
c. $\quad 1 \mathrm{mg} / \mathrm{L}$
d. $1 \mathrm{~kg} / \mathrm{L}$
e. None of the above are correct.

## PHARMACIST LEARNING OBJECTIVES

1. Identify the components of halitosis and potential treatment options for the patient
2. Recognize the importance of fluoride in dental hygiene

## PHARMACY TECHNICIAN LEARNING OBJECTIVES

1. Identify the components of halitosis and potential treatment options for the patient
2. Recognize the importance of fluoride in dental hygiene

## OVERVIEW

Micro-learning opportunities were created in response to evidence that learning is maximized when delivered in short and focused 'bursts.' In this session, halitosis is described and different mouthwash therapies are explained.

## ACCREDITATION



PharmCon is accredited by the Accreditation Council for Pharmacy Education as a provider of continuing pharmacy education.

PharmCon reports CPE credits to CPE Monitor automatically after credit is earned. Your NABP ePID and birth date must be in your online profile for successful credit submission.

PharmCon reports CPE credits to CE Broker automatically after credit is earned. Your license number must be in your online professional profile for successful credit submission.

PharmCon is approved by the California Board of Registered Nursing (Provider Number CEP 13649) and the Florida Board of Nursing (Provider Number 50-3515). Activities approved by the CA BRN and the FL BN are accepted by most State Boards of Nursing.

ACPE accredited programs provided by PharmCon meet requirements for American Nurses Credentialing Center (ANCC) Category 1 pharmacotherapeutic/pharmacology credit towards
certification renewal. ACPE accredited programs are listed by the ANCC and AANP as an acceptable, accredited continuing education organization for applicants seeking renewal through continuing education credit.

## AUTHOR DISCLOSURE

Peter Kreckel reports no financial relationship with the manufacturer(s) or provider(s) of any commercial interest(s) or service(s) that appear in this program.

ACTIVITY TYPE: Knowledge-based
RELEASE DATE: 10/17/2023
EXPIRATION DATE: 04/17/2026

## ACPE UNIVERSAL ACTIVITY NUMBER

PHARMACIST: 0798-0000-23-287-H01-P
PHARMACY TECHNICIAN: 0798-0000-23-287-H01-T
CE BROKER ID NUMBER: 20-1117624
EDUCATIONAL SUPPORT PROVIDED BY: PharmCon

## TARGET AUDIENCE

Pharmacist, Pharmacy Technician

